



# EFFCET OF NANOEDIBLE COATING OF FRENCH FRIED POTATOES AND OIL UP TAKE REDUCTION

<sup>1</sup>Sanaa M. Ali, <sup>2</sup>WafaaBayoumy, <sup>2</sup>M.Khairy, <sup>1</sup>Manal A. Sorour, <sup>2</sup>M.A. Mousa

<sup>1</sup>Food Engineering and Packaging Department, Food Technology Research Institute, ARC

<sup>2</sup>Chemistry Department, Faculty of science, Benha University, (Egypt)

## ABSTRACT

*The effect of nanoparticles of chitosan and carrageenan as an active coating on oil reduction of potato strips during frying was investigated using different concentrations of chitosan and carrageenan. The rheological properties of edible coating materials showed non –Newtonian pseudo plastic behavior. The results illustrated that nano particle of carrageenan (24nm) reduced oil absorption from 12% to 4.93% at 170°C and moisture content was (54.83%).*

## I INTRODUCTION

The potato products such as French fries, chips, balls, etc. are also fried products which absorb a high amount of oil. These products though taste right, pose health problems due to high calorie and cholesterol intake. [1]

An edible coating is a thin layer of edible material formed as a coating on a food product, [2] While an edible film is a preformed thin layer, made of the edible material, which can be placed on or between food components [3,4].

The main difference between these two food systems are that the edible coating is applied in liquid form on the food. This is usually done by immersing the product in the solution of edible material, the edible film is first molded as solid sheets, then applied as a wrapping for food products [5]. Their growing application is attributable to reduction of moisture loss, adverse chemical reactions [6,7]. Spoilage, and microbial contamination [8]. Additionally, they can be used for controlled release of food additives [9]. Edible coatings are also effective as a post-harvest treatment to preserve fruit quality [10].

Rheology is the branch of science dealing with the flow and deformation of materials. Rheological instrumentation and rheological measurements have become essential tools in the analytical laboratories for characterizing component materials and final products, monitoring process conditions, as well as predicting product performance and consumer acceptance [11].

The flow behavior of the edible coating solutions needs to be studied since the viscosity of the film forming solution is key to controlling the desirable thickness of the coating [12]. The viscosity of the coating solutions is also important for decreasing the dewetting process which prevents the creation of a continuous layer around food, making it necessary that the magnitude of the viscous forces be greater than that of the interfacial ones [13]. Edible films are



made of various materials, are formed by different processes, and have various properties. Edible film formers include polysaccharides, proteins, and lipids. Polysaccharides may include cellulose derivatives; starches and their derivatives; seaweed extracts such as carrageenan and alginates; pectin and chitosan. Protein film formers include collagen, gelatin, whey protein, corn, zein, soy protein, and wheat gluten. Polysaccharide and Protein film materials are characterized by high moisture permeability, low oxygen and lipid permeability at lower relative humidities, as well as a compromised barrier and mechanical properties at high relative humidities [14].

Deep-fat frying is a widely used method for preparing foods with an attractive and tasty surface. The soft and moist interior along with the porous outer crispy crust provides increased palatability to foods [15].

One of the critical quality attributes of deep fat fried products is the amount of oil content in these products. Fried food with low-fat content, can hard texture edible coating have long been known to protect perishable food, from deterioration by retarding dehydration suppressing respiration, improving texture quality, helping to retain volatile flavor compounds and reducing microbial growth. Another application of edible films of the coating is a barrier to lipid absorption by food during deep fat frying. Oil uptake in fried foods has become a health concern, high consumption of lipids been related to obesity and other health problem such as coronary heart disease [16].

The aim of the work is to study the effect of different particle size of nanoparticle solutions as coating material compared with the various concentrations of chitosan and carrageenan solution to use the coating as mentioned above materials to reduce fat absorption during the frying process. Also, the rheological properties of the corresponding edible coating mixtures have been studied with the aim of elucidating the phase behavior and measuring the viscoelastic properties of the mixtures on the way to the film formation.

## II MATERIAL AND METHOD

### 2.1. Materials

Chitosan and Sodium tripolyphosphate (TPP) were purchased from Acemtic Company. Carrageenan was purchased from Mefad Company.

### 2.2. Methods

#### 2.2.1. Preparation of potato strips

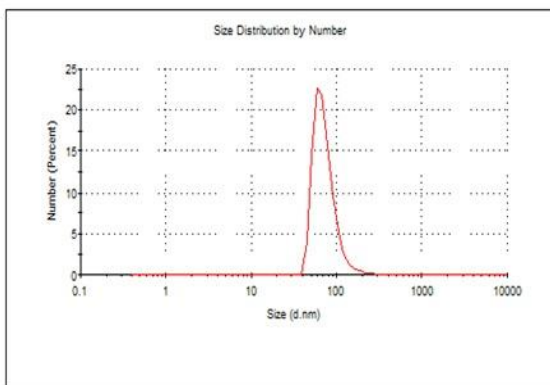
Potato tubers were washed, hand-peeled and cut with a manual operated potato-cutting device into an 8×8×60 mm strips, then rinsed in tap water.

#### 2.2.2. Preparation of Chitosan nanoparticles

Chitosan nanoparticles were prepared according to [17]. Chitosan was dissolved in (1%) acetic acid with stirring until to be completely soluble. The TPP solution was added to chitosan solution drops with different chitosan: TPP ratios (1:1, 2:1 and 3:1).

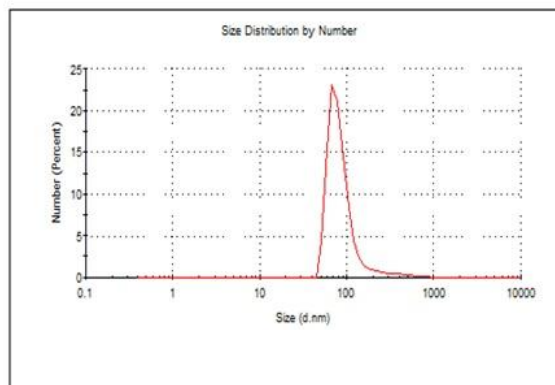
Results

	Size (d.nm)	% Number	Width (d.nm)
Peak 1:	75.99	100.0	47.66
Peak 2:	0.000	0.0	0.000
Peak 3:	0.000	0.0	0.000



Results

	Size (d.nm)	% Number	Width (d.nm)
Peak 1:	93.19	100.0	70.91
Peak 2:	0.000	0.0	0.000
Peak 3:	0.000	0.0	0.000



Results

	Size (d.nm)	% Number	Width (d.nm)
Peak 1:	151.1	86.5	37.11
Peak 2:	905.6	13.2	281.8
Peak 3:	0.000	0.0	0.000

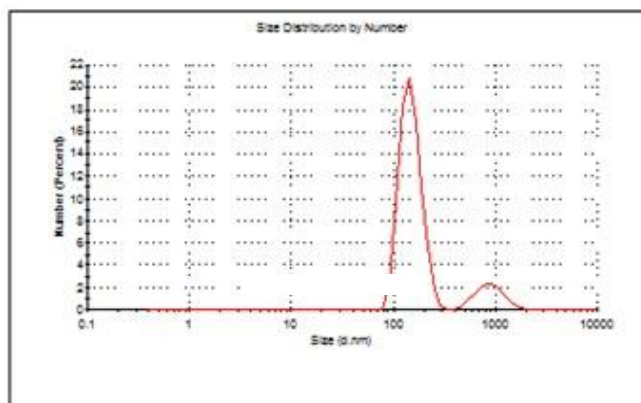


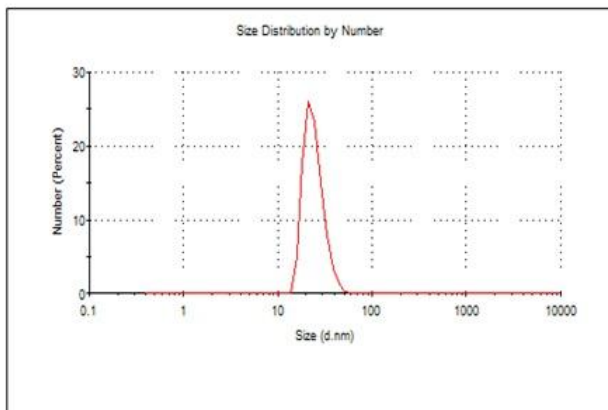
Fig.(1) Distribution particle size of nano chitosan

### 2.2.3. Preparation of nano Carrageenan particles

Carrageenan nanoparticles were prepared by dissolving carrageenan in distilled water with stirring at 70°C. The TPP solution was added to carrageenan solution drop wise at carrageenan: TPP ratio (1:1).

Results

	Size (d.nm)	% Number	Width (d.nm)
Peak 1:	24.05	100.0	6.024
Peak 2:	123.5	0.0	34.47
Peak 3:	0.000	0.0	0.000



Results

	Size (d.nm)	% Number	Width (d.nm)
Peak 1:	35.74	100.0	5.127
Peak 2:	0.000	0.0	0.000
Peak 3:	0.000	0.0	0.000

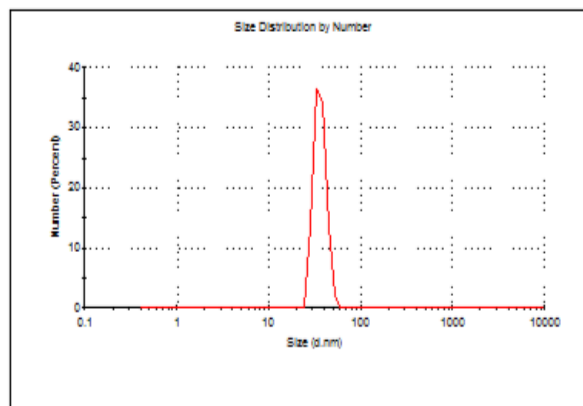


Fig. (2) Distribution Particle size of carrageenan

#### 2.2.4. Rheological Properties of edible coating solution

Rheological properties (viscosity, shear stress, shear rate) of different concentrations of nano-chitosan and nano-carrageenan solutions were measured using Brookfield Engineering Labs DVIII Ultra Rheometer. The samples were placed in a small sample adapter at a constant temperature using thermostatic water bath. The viscometer was operated between 10 and 100 rpm, and the results of shear stress, shear rate, and viscosity data were obtained directly from the instrument; the SC4-21 spindle was selected for the measurement.

#### 2.2.5. Preparation of potato strips before frying:

Three lots of potato pieces were separately immersed at room temperature for two minutes in coating solutions of a) different concentrations of chitosan, b) various concentrations of carrageenan, c) different particle size solution of nano-chitosan and d) different particle size of nano carrageenan. The samples were removed and then blotted with filter paper to remove surface moisture. Sun flower oil was used as a frying medium, an amini fryer with 1 L capacity was used for frying operation. The samples were immersed in the hot oil (170 and 190°C) and fried for 4-5 min. Fried samples were removed from the unit and the excess surface oil absorbed with filter paper. Samples were then allowed to cool to room temperature for 5 min before oil content analysis was done. The oil and moisture contents were determined using Soxhlet extraction method and oven drying method at 105°C until constant weight respectively according to the guidelines proposed by [18].

### 2.2.6. Analysis of the Coating Material

Yield parameters were determined by measuring the mass of the raw potato strips (X), the mass of the coated potato strips before frying (Y) and the mass of the coated potato strips after frying(Z). The yield parameters were calculated using the following equations [19].

$$\text{Adhesion degree} = \frac{Y-X}{Y} \times 100 \quad (1)$$

$$\text{Yield} = \frac{Z}{X} \times 100 \quad (2)$$

$$\text{Frying loss} = \frac{Y-Z}{Y} \times 100 \quad (3)$$

## III RESULTS AND DISCUSSION

### 3.1.Rheological properties of chitosan and carrageenan solution

Rheological properties of chitosan and carrageenan solution were measured for the prepared blends to investigate the flow behavior of blends which is an important factor for food coating materials. Accurate rheological characterization is of critical importance regard in terms of commercial processibility, physiological activity product design,...etc. [20].

Figure(3) shows the effect of shear rate on the apparent viscosity of different concentrations of chitosan solution. The results indicated that by increasing the concentration of chitosan, the apparent viscosity increased at all shear rates studied. More pseudo plastic is observed with increasing chitosan concentration, this can be explained concerning the degree of chain entanglements, as polymer concentration is increased, the freedom of movement of the individual chains becomes restricted due to the correspondingly increased number of complications.

The shear thinning region curve can be expressed by the following power law equation.

$$\mu = K\gamma^{n-1}$$

Where  $\mu$  is the apparent viscosity (Pa.s), K is the consistency index,  $\gamma$  is the shear rate (1/ s), n is the flow behavior index.

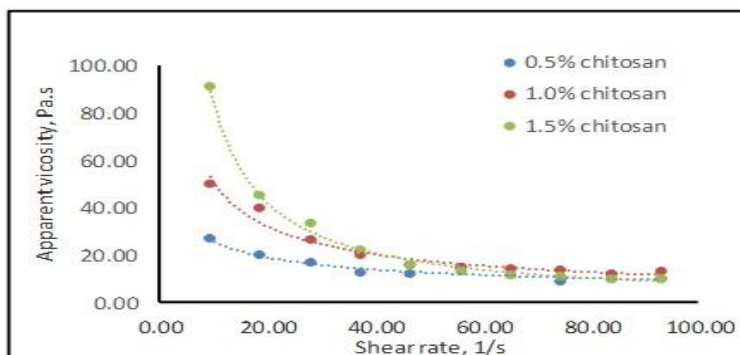


Fig.(3) Effect of shear rate on apparent viscosity of chitosan solution

Figure (4) illustrates the effect of shear rate on apparent viscosity of different concentration of carrageenan solution (0.5, 1 and 1.5%). The results indicated that apparent viscosity decreased as shear rate increased for all concentration studied, as previously discussed by [21]. The shearing behavior of carrageenan solution was found to be fitted by power law equation.

The low concentration of carrageenan solution show low pseudo plasticity since the rate of disentanglements is higher than that entanglement.

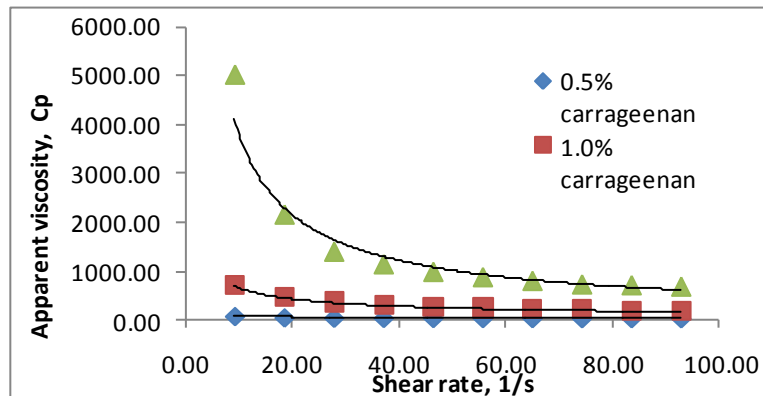


Fig.( 4 ) Effect of shear rate on apparent viscosity of Carrageenan solution

### 3.2. Rheological properties of nanoparticle suspended solution.

Figure (5) displays the relation between shear rate and apparent viscosity for chitosan and corresponding chitosan/tripolyphosphate solutions. The results indicate that the suspensions exhibited non-Newtonian pseudo plastic behavior since the viscosity decreased with increasing the shear rate. These results are following the findings of [22]. Apparent viscosity decreased as particle size of nano-chitosan increased, and this may be due to that the larger the particle size, the higher will be the resistance offered for flow of liquid and hence the higher will be the viscosity and vice versa [23].

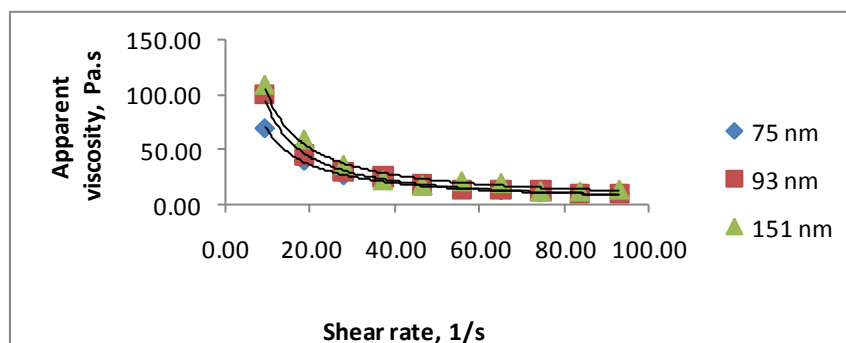


Fig.(5) Effect of shear rate on apparent viscosity of chitosan nanoparticles

Figure(6) shows the effect of shear rate on apparent viscosity of nano-carrageenan suspended solution. The results indicated that apparent viscosity increased as increasing the particle size of nano-carrageenan and decreased as shear rate increased.

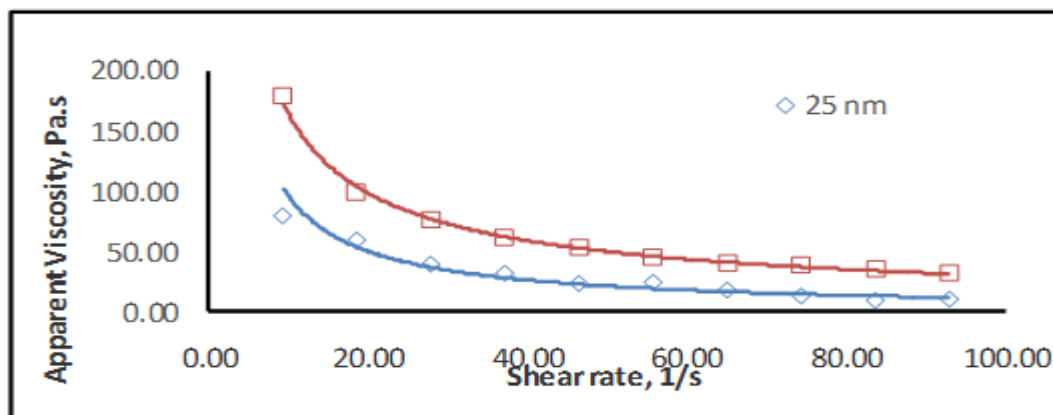


Fig. (6) Effect of shear rate on apparent viscosity of carrageenan nanoparticles

### 3.3. Effect of edible coating on adhesion degree, yield and frying loss

Table (1) shows the effect of different edible coatings on adhesion degree, which was determined using equation(1).The results obtained are listed in Table (1). It showed that the highest degree for adhesion (32.966 %) was foundfor potato strips coated with nano-carrageenan (24nm), while the lowest adhesion degree was recorded for samples coated with 0.5% chitosan (1.379%).

The effect of different edible coatings on yield percent was also studied and determined using equation (2) and given in Table ( 1 ).The results showed that the highest yield % (97,227%) was found for samples coated with nano carrageenan(24nm) with ,whereas the lowest yield percent was recorded for samples coated with nano chitosan (150nm).

The effect of different edible coatings on frying loss was determined using equation (3). The results illustrated that the highest frying loss (15.79%) was found for samples coated with 0.5% chitosan and the lowest frying loss was recorded(9.45%) for samples coated with nano carrageenan(24nm).

### 3.4. Effect of edible coating on moisture content

The edible coated samples retained more moisture in the surface layer than control samples. Table (1) shows the moisture content of the sample after final frying. The results indicated that using edible coating for frying of potato strips caused increasing in moisture contents of the products. Nano-carrageenan (24nm) showed that the best results with respect to the edible efficiency coating in increasing moisture content during frying (170°C) process of potato strips. Chitosan (0.5%) had lower moisture content, while it increased as concentration of chitosan increased. Moisture content decreased as particle size of nanochitosan increased, this may be due to that coating was

proportional to the increase in concentration of edible coating, indicating greater moisture retention by thicker films[21].The results illustrated that moisture content increased with increasing frying temperature, This may be due to that increasing the frying temperature made harder and more durable coating, which could be increased the resistance of products against mobility of the materials and improved a surface diffusivity. In the present study increased frying temperature retains more moisture, [23].

**3.5. Effect of edible coating on oil absorption**

The reduction of oil content of coated and uncoated potato strips are shown in table (1). A reduction of oil up take was observed for deep fat fried coated potato strips samples compared to control samples. The surface layer coated sample absorbed less oil compared to the control sample, edible films as gel-forming compounds forms a fine net structure which prevents the oil migration in the potato tissue during the frying process according to the type of film used[24].The results show that fat rate for potato strips coated with suspended nanoparticle solutions in comparison with control samples and that which causes the oil to diffuse in a counter direction (from inside to outside of the food), this is in agreement with (Diaze, et al., 1999) Among all edible coating tested, nano-carrageenan (24nm) and nano-chitosan (75nm) provided that the lowest fat content are 4.933 and 5.09%, respectively. This may be due to the fact that potato strips coated with nanoparticles suspensions causes the oil to diffuse in counter direction from inside to outside of the food [25,26].

**Table1. The effect of coating materials on the values of adhesion degree, yield, frying loss, Moisture, and Fat percent**

Sample	Adhesion degree, %	Yield, %	Frying loss	Fat%		Moisture content	
				170°C	190°C	170°C	190°C
Control	---	---	---	13.23	12.23	52	53
0.5% Chitosan	1.379	85.384	15.793	8.6	10.928	53.846	55.1
1% Chitosan	6.962	91.496	14.873	7.133	8.45	57.213	55.22
1.5% Chitosan	7.82	95.685	10.068	7.4	7.352	58.745	58.46
75nm nano-chitosan	25.376	86.704	11.561	5.09	5.07	67.53	61.53
93nm nano-chitosan	22.785	76.972	16.58	5.333	6.05	64.45	65
150nm nano-chitosan	11.8368	65.876	22.881	6.866	7.125	54.55	66.55
0.5% Carrageenan	9.87	88.34	20.046	11.33	9.78	65.1	58.34
1% Carrageenan	14.062	93.036	19.8	9.857	9	65.22	64.213
1.5% Carrageenan	22.22	95.897	16.301	8.43	7.2142	68.46	65.745
nano carrageenan 25nm	32.966	97.227	9.45	4.933	5.533	68.83	67.83
nanocarrageenan 34 nm	19.45	83.554	12.64	7.933	5.733	63.53	65.53



#### **IV CONCLUSION**

Chitosan and Carrageenan nano particles exhibited non-Newtonian pseudoplastic behavior since the apparent viscosity of edible coating solutions decreased as shear rate increased. The effect of nano particle solutions of chitosan and carrageenan on oil uptake reduction during frying was investigated and compared with that of bulk size of chitosan and carrageenan.

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